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In chemistry, a coordination complex consists of a central atom or ion, which is usually metallic and is called the coordination centre, and a surrounding array of bound molecules or ions, that are in turn known as ligands or complexing agents. Many metal-containing compounds, especially those of transition metals, are coordination complexes. A coordination complex whose centre is a metal atom ...

### Coordination complex - Wikipedia

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The group 11 metals are typically categorised as transition metals given they can form ions with incomplete d-shells. Physically, they have the relatively low melting points and high electronegativity values associated with post-transition metals.

### Post-transition metal - Wikipedia

Les métaux alcalins sont les éléments chimiques de la première colonne du tableau périodique des éléments [19], sauf un. Le lithium (Li), le sodium (Na), le potassium (K), le rubidium (Rb), le césium (Cs) et le francium (Fr) font ainsi partie de cette famille, mais pas l'hydrogène [20]. Aucun des éléments du groupe n'existe à l'état d'élément pur dans la nature, on ne les retrouve ...

### Chimie minérale â€” Wikipedia

Als Supersäuren werden in der Chemie Säuren bezeichnet, die stärker als konzentrierte (100-prozentige) Schwefelsäure ( $H_2SO_4$ :  $pK_s$ -Wert =  $\hat{a}^{\sim}3,0$ ) sind.. Alle Supersäuren haben somit einen  $pK_s$ -Wert im negativen Bereich. Zur Quantifizierung der Säurestärke wird die Hammettsche Aciditätsfunktion verwendet.. Fluorsulfonsäure ( $HSO_3F$ ) ist beispielsweise mehrere tausendmal stärker als ...

### Supersäuren â€” Wikipedia

Metallcarbonyle sind Komplexverbindungen von Übergangsmetallen mit Kohlenstoffmonoxid als Ligand. Die Metalle treten in diesen Verbindungen mit der Oxidationszahl  $\hat{A}\pm 0$  auf und weichen in ihren Eigenschaften zum Teil erheblich von den klassischen Metallverbindungen ab. Einkernige Metallcarbonyle mit nur einem Zentralatom sind oft gut in organischen Lösungsmitteln löslich und leichtflüchtig.

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